

EXTRAPOLATION OF AN ISENTROPE OF A SOLID IN THE REGION OF HIGH PRESSURES

V. A. Filimonov

UDC 539.89

The behavior of a material under compression is determined by its equation of state, which relates three (or more) thermodynamic parameters; for example, the pressure  $p$ , the temperature  $T$ , and the specific volume  $v$ . We consider barotropic processes for which the pressure is a single-valued function of the volume. In the general case of compression, the stress tensor is nonspherical; we shall, however, consider only states of isotropic (hydrostatic) compression in which the stresses are characterized only by the scalar  $p$ . Information concerning the behavior of a material under compression is gained from experiments involving isothermal (static) compression as well as from shock-wave measurements. At the present time, an isotherm for most materials is determined experimentally over a fairly narrow interval,  $\Delta v/v_0 = 0.2$  to  $0.1$ , of compression. For many material the shock adiabat  $p_s(v)$  is known for  $v_0/v$  in a range from two to three. In an interval of large compressions the equation for an isotherm or for the cold pressure (the zero isotherm) is obtained in a variety of ways: 1) The cold pressure is deduced from the shock adiabat upon making certain loosely justified assumptions concerning the behavior of the equation of state [1-3]; 2) the isotherms of Birch and Keene [4] are obtained from various model representations; 3) the cold pressure is obtained theoretically with the aid of the method of the pseudopotential (up to the present time this technique has been successful only for the alkali metals [5-7]); 4) an equation for the pressure is obtained by extrapolating experimental data without drawing on additional physical assumptions concerning the behavior of the equation of state of the material.

We consider the last method. As the basic parameter defining the compressibility of a material, we have the modulus of compression

$$K_v(v) = -v(\partial p/\partial v)_v$$

( $v$  is the index of barotropicity), the initial value for which is known for many materials. As the simplest extrapolation of a barotrope we have the expression

$$p = (K_{0v}/v_0)(v_0 - v), \quad p(v_0) = 0, \quad K_{0v} = K_v(v_0),$$

which represents the first term in the Taylor-series expansion for the pressure in terms of the volume. A more exact extrapolation can be had if one knows an experimental value for the derivative of the modulus of compression:

$$K_p = (\partial K_v/\partial p)_v.$$

Then, putting  $K_p = \text{const}$  [4, 8], we obtain the equation

$$p = (K_{0v}/K_p) [(v_0/v)^{K_p} - 1], \tag{1}$$

which in the isothermal case is referred to as Murnaghan's equation. For a more exact extrapolation of a barotrope, experimental data are currently augmented by means of certain model representations, for example, by a functional form of the interatomic interaction potential [4]. A number of ways exist for describing barotropic equations. However, for practical use (for example, to numerically solve gasdynamic problems), it is desirable to have a uniform form for the equation of a barotrope. In order to obtain uniform expressions for the barotrope of various materials, extrapolating them more exactly than does (1), we introduce the auxiliary function

$$B_v(v) = (K_p + 1)/4 = -(v/4)p''/p' \tag{2}$$

(the primes indicate differentiation with respect to volume). In terms of the function  $B_v$ , we define the zero isotherm  $B_C$ , the isotherm  $B_T$ , the isentrope  $B_S$ , and the shock adiabat  $B_S$ . We calculate the pressure from (2) as follows:

---

Novosibirsk. Translated from Zhurnal Prikladnoi Mekhaniki i Tekhnicheskoi Fiziki, No. 6, pp. 105-111, November-December, 1978. Original article submitted November 30, 1977.

$$p(v) = p_0 + \frac{K_{0v}}{v_0} \int_v^{v_0} \exp\left(4 \int_w^{v_0} B_v(x)/x dx\right) dw. \quad (3)$$

We now establish some properties of the various barotropes  $B_v$ . For qualitative estimates we use the Mie-Grüneisen equation of state,

$$p(v) = p_c(v) + (\gamma(v)/v)(E - E_c), \quad (4)$$

where  $E$  is the specific internal energy;  $E_c$  is the elastic energy;  $p_c$  is the cold (elastic) pressure; and  $\gamma(v) \approx 2$  is the Grüneisen coefficient, assumed to be a weakly varying quantity [9]. For  $\tau = (p - p_c)/K_{0c} \ll 1$ , from (4) we obtain

$$B_S(v) = B_c(v) + A_S(v)\tau, \quad (5)$$

where  $A_S$  can be obtained when the function  $\gamma(v)$  is known. We write the equation of an isotherm in the form

$$p_T = p_c + \frac{\gamma(v)}{v} \int_0^T C_v dT,$$

from whence we obtain

$$B_T(v) = B_c(v) + A_T(v)\tau, \quad (6)$$

where  $A_T(v)$  depends on  $\gamma(v)$  and the heat capacity  $C_v(v)$ .

The function  $B_S(v)$  has a fairly simple physical interpretation [8], namely, the following: At the initial point of the shock adiabat

$$B_S(v_0) = B_s(v_0) = dU/du, \quad (7)$$

where  $U$  is the shock-wave velocity and  $u$  is the particle velocity. The use of the functions  $B_S(v)$  makes it possible to write the equations of one-dimensional isentropic gasdynamics in the form of the characteristics and the relationships on them in the following way:

$$\begin{aligned} dx/dt &= u \pm c, \quad d(u \pm c)/du = 2B_S(v) + 1, \\ c &= c_0 \exp\left(2 \int_v^{v_0} B_S(w)/w dw\right), \end{aligned} \quad (8)$$

where  $x$  is a coordinate,  $t$  is the time, and  $c$  is the sound velocity. The empirical relationship

$$U = a + bu, \quad (9)$$

which approximates well the results of shock-wave measurements [10] ( $a$  and  $b$  are constants), allows us to estimate the initial value of an isentrope through use of (7); thus,

$$B_S(v_0) \simeq b = 1-2.$$

The sign of approximate equality reflects the fact that the relationship (9) applies for high pressures (high yield strength) when the compression can be considered to be isotropic.

In the limiting case  $v \rightarrow 0$ , a material can be considered to be a degenerate Fermi gas [11],

$$p(v) \sim v^{-5/3},$$

from which we have

$$\lim_{v \rightarrow 0} B_{c,T,S} = 2/3.$$

The properties of the functions  $B_v(v)$ , considered here, enable us to assume that in the absence of phase transitions (jumps in  $p'$  and  $p''$ ), this function is bounded,  $B \sim 1$ . It is therefore convenient to obtain the barotrope equation from (3), where  $B(v)$  is approximated by the Taylor series

$$B(v) = B_0 + B_1(v - v_0) + B_2(v - v_0)^2 + \dots \quad (10)$$

It is desirable to know how errors made in determining  $B(v)$  affect the determination of  $p(v)$ .

The expression (3) allows us to find the relationship between the variations  $\delta B$  and  $\delta p$ . Thus, we have

$$|\delta p|/p(v) \leq \left| \exp \left( 4 \int_v^{v_0} (\delta B/w) dw \right) - 1 \right|;$$

for  $(v_0 - v)/v_0 \ll 1$  and  $\delta B = \Delta = \text{const}$ , we obtain

$$|\delta p|/p \leq 4|\Delta|(v_0 - v)/v_0,$$

while for the variation  $|\delta B| = H_n(v - v_0)^n$ , by which we can estimate the remainder of the series (10) with  $n$  first terms, we obtain the expression

$$|\delta p|/p = [4H_n/(n+1)](v_0 - v)v_0^{n+1}.$$

We use experimental data to determine the coefficients of the series (10). Using data for an isothermal compression [12-15], we write the function  $B_T(v)$  in the form  $B_T = B_0 + B_1(v - v_0)$ . We select the coefficients  $B_0$  and  $B_1$  from the condition of best approximation to the experimental data:

$$\sigma^2 = [\min(B_0, B_1)] \frac{1}{q} \sum_{i=1}^q (1 - p(v_i, B_0, B_1)/p_e(v_i))^2,$$

where  $p_e(v)$  is the pressure, obtained experimentally;  $p(v, B_0, B_1)$  is the pressure calculated from (3). Results from the calculations are presented in Table 1.

We note that for Na the values of  $B_0$ , calculated with the aid of the data of the various authors, are close to one another. At the same time, the values of  $B_1$  differ sharply from one another (see Table 1), which forces one to doubt the possibility of calculating  $B_1$  in the remaining cases also. Consequently, to approximate the data from an isothermal compression, it is sufficient in (10) to take into account the first term; i.e., for the pressure we obtain an equation of the form (1). Taking the second term into account does improve the approximation to the experimental data but it does not add to our knowledge concerning the true behavior of  $p(v)$ . We return now to the results of the shock-wave measurements and we attempt to obtain from them the successive coefficients in (10). To do this, we establish relationships for the derivatives of  $p_s$  and  $p_S$  with respect to the volume at the initial point of the shock adiabat,  $v = v_0$ ,  $p_s = p_S = p_0$ ,  $E_s = E_S = E_0$  ( $E$  is the specific energy).

We write the energy-balance equation upon passage through the shock. Thus, we have

$$E_s - E_0 = (p_s + p_0)(v_0 - v)/2. \quad (11)$$

We write the difference between  $p_s$  and  $p_S$  in the series form

$$p_s - p_S = \alpha(E_s - E_S) + \beta(E_s - E_S)^2 + \dots, \quad (12)$$

where

$$\alpha(v) = (\partial p / \partial E)_v; \quad \beta(v) = (\partial^2 p / \partial E^2)_v / 2.$$

The first term of the series corresponds to the Mie-Grüneisen equation of state (2), where  $\alpha(v) = \gamma(v)/v$ . Differentiating (11) and (12), we obtain the first five derivatives, which do not contain  $\beta$ :

$$\begin{aligned} p'_s &= p'_S, & p''_s &= p''_S, & p'''_s &= p'''_S - (\alpha/2) p''_S, \\ p_s^{(IV)} &= p_S^{(IV)} + (\alpha^2/2 - 2\alpha') p''_S - \alpha p_S''', \\ p_s^{(V)} &= p_S^{(V)} - (3/2) \alpha p_S^{(IV)} + [(3/2) \alpha^2 - 5\alpha'] p_S''' - [5\alpha'' + \alpha\alpha'/2 + (3/4) \alpha^3] p_S''. \end{aligned} \quad (13)$$

The first two equations are trivial and yield the known result (7),

$$B_s = B_S.$$

For the first derivative of  $B_S$  we obtain

$$B'_S = B_1 = B'_s + (\gamma/2v) B_s, \quad (14)$$

where the value of  $\alpha$  is replaced by  $\gamma/v$ . The relation (14) allows us to obtain  $B_1$  from the experimentally known adiabat  $p_s(v)$ ; for this it is sufficient to know the initial value of the Grüneisen coefficient, which for the majority of materials is determined experimentally [3].

TABLE 1.

Case	Material investigated	T, °K	$v_0'$ cm <sup>3</sup> /gm	$K_0$ , kbar	$B_0$	$v_0 B_1$	Calc. value	Exptl. value
							σ, %	
1	Na [12]	400	1,086	60,82	1,21	0*	1,2	0,2
2		400	1,086	60,82	1,28	0,67	0,1	0,2
3	[13]	201	1,010	64,56	1,16	-0,13	1,3	0,2
4		20	0,986	74,91	1,09	-0,13	2,0	0,2
5	K [14]	0	1,108	37,00	1,22	0*	0,7	0,5
6		0	1,108	37,00	1,25	-0,31	0,4	0,5
7	Ca [15]	0	0,645	182,63	0,99	0,43	0,5	—
8	Ba [15]	0	0,265	94,62	0,78	0,13	1,0	—

\*The minimization of  $\sigma$  in (13) was carried out only with respect to  $B_0$ ; the value of  $B_1$  was set equal to zero.

We can obtain the function  $B_S(v)$  through use of the empirical relation (9), from which we get an expression for the pressure all over again [8], and then we have

$$B_S(v) = bvy[1 + (3/2)b(v_0 - v)y]/[1 + 2b(v_0 - v)y], \quad (15)$$

where  $y = [v_0 - b(v_0 - v)]^{-1}$ . The expression (15) describes  $B_S(v)$  for  $v < v^* < v_0$ ; i.e., both it and (9) are suitable expressions over this interval. In Fig. 1 we show the function  $B_S(v/v_0)$  for Na (curve 1), where the solid curve corresponds to (15) for  $b = 1.26$  (see [10]) and the dashed curve represents the assumed behavior in a neighborhood of  $v/v_0 = 1$  ( $v_0 = 1.086$  cm<sup>3</sup>/g,  $B_0 = 1.2$ ). From Fig. 1 it is evident that  $B_S' \ll 1$ , and, therefore, upon neglecting this quantity, we obtain

$$v_0 B_1 = \gamma B_0/2 \simeq 1$$

( $\gamma = 1.8$  from [8]). The line segment 2 in Fig. 1 corresponds to a linear approximation for  $B_S$ ; the line segments 3, 4, and 5 correspond to a linear approximation for  $B_T$  in accordance with the tabular data (data in rows 2, 3, and 4). We conclude, therefore, that the available experimental data is sufficient for extrapolating  $B_S$  linearly with respect to the volume.

To improve the extrapolation of  $B_S(v)$ , it is desirable to know the following terms of the series (10). Using (13), we can obtain  $B_S''$  and  $B_S'''$ , independently of  $\beta(v)$ . Thus, for  $B_S$  we obtain an expression which contains  $\gamma_0'$ :

$$B_S'' = 2B_2 = B_S'' + \alpha B_S' - 2\alpha' B + 2\alpha B^2/v.$$

An analogous expression can be obtained for  $B_S'''$ , but involving  $\gamma_0''$ . At the present time, experimental data [10] do not permit even a crude determination of  $\gamma_0'$  and  $\gamma_0''$ .

Approximate theoretical expressions are available which relate  $\gamma(v)$  and the cold pressure  $p_c(v)$ . These expressions, obtained by various authors, can be subsumed in a common form (see [3]), namely,

$$\gamma(v) = -(v/2)(p_x v^n)''/(p_x v^n)' + \mu, \quad (16)$$

where  $n$  and  $\mu$  are constants; in particular, the values  $n = 0$  and  $n = 2/3$  yield, respectively, the formulas due to Landau-Slater and to Dugdale-MacDonald. Expressions for the cold pressure  $p_c(v)$  of certain materials, obtained with the use of (16) and Eq. (4), have appeared in [1-3].

Differentiating (16), we have

$$v_0 \gamma_0' = n(2B - 5/2 - n/2) + 2B_S' v_0. \quad (17)$$

A similar procedure yields an expression for  $\gamma_0''$ . Factors such as the thermal pressure of the electrons, the heat capacity, the anharmonicity of the lattice vibrations, and others enter into the second term of (12) and the fifth term of (10).

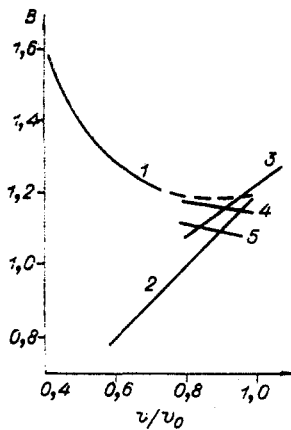


Fig. 1

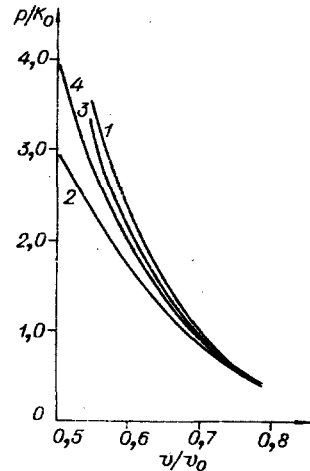


Fig. 2

At the present time, the approximate formula (16), wherein differing  $n$  values yield differing values of the initial derivative  $\gamma_0$ , [see the relation (17)], exhausts our knowledge of how the Grüneisen coefficient depends on the volume. Inaccuracy in the determination of  $\gamma(v)$  begins to manifest itself in the third term of (10); therefore, to take into account a fifth term hardly leads to any refinement in the function  $B_S(v)$ . To obtain the zero isentrope (cold pressure) by means of our method, it is necessary to know the shock adiabat of the material starting from a temperature close to absolute zero. For moderate temperatures we can neglect differences among  $B_S$ ,  $B_T$ , and  $B_C$  [see the relations (5) and (6)] and use the resulting  $B_S$  to calculate the isothermal and the cold pressure in accordance with (3). To illustrate our method, in Fig. 2 we display  $p_S(v)$  for Cu, wherein the curves 1, 2, and 3 correspond, respectively, to taking into account one, two, and three terms of the series (10); curve 4 corresponds to the cold pressure  $p_C(v)$  from [2].

Uniformity of the barotrope equation is understood here in the sense of uniformity in the representation of  $B_0(v)$  by means of the power series (10). From the latter we can obtain the function  $p(v)$  in the necessary form and with the necessary accuracy. For example, using two terms of (10),  $B(v) = B_0 + B_1(v - v_0)$ , we can, with accuracy to  $O(v - v_0)^3$ , put

$$B(v)/v = B_0^2 / (B_0 + (B_0 - v_0 B_1)(v - v_0)),$$

from which we get

$$p(v) = [K_0 / (4\beta - 1)] [((\omega + v/\rho) / (\omega + v_0/\rho))^{4\beta-1} - 1],$$

where

$$\omega = v_0^2 B_1 / B_0^2; \quad \rho = B_0^2 / (B_0 - B_1 v_0).$$

In the general case,  $B(v)$  may be extrapolated by means of several terms of the series (10). The accuracy of such an extrapolation depends on the accuracy of the experimental data used and also on the magnitude of the remainder of the series (10). The extrapolation of  $B(v)$  can be improved if experimental values of  $B(v)$  are available for several values of  $v$ . To acquire such values, we can suggest two types of experiments:

1. Measurement of the shock-wave velocity  $U$  and the particle velocity  $u$  in a precompressed (statically or dynamically) material. Then  $B_S(v) = dU/du$  [see the relation (7)].

2. Realization of a simple wave experimentally (the wave being one of either compression or rarefaction). For an isentropic one-dimensional flow we obtain, from the relation along the characteristics (8), the result

$$\frac{\partial}{\partial t} (\partial x / \partial u)_t = \pm (2B_S + 1). \quad (18)$$

By measuring  $u(x)$  at two different times, we obtain  $B_S(u)$  from (18). For determining  $v(u)$  we have the differential equation

$$d^2 v / du^2 = \pm (2B_S(u) + 1)(dv/du)/v,$$

which can be solved numerically for arbitrary  $B_S(u)$ .

Questions as to the feasibility of such measurements call for a separate study.

The author expresses his thanks to Yu. I. Fadeenko for his statement of the problem and for valuable discussions.

#### LITERATURE CITED

1. R. G. McQueen and S. P. Marsh, "Equation of state for nineteen metallic elements from shock-wave measurements to two megabars," *J. Appl. Phys.*, 31, No. 7, 1253 (1960).
2. L. A. Al'tshuler, S. B. Kormer, A. A. Bakanova, and R. F. Trunin, "Equation of state of aluminum, copper, and lead for a region of high pressures," *Zh. Eksp. Teor. Fiz.*, 38, No. 3, 720 (1960).
3. V. N. Zharkov and V. A. Kalinin, *Equations of State of Solids at High Pressures and Temperatures* [in Russian], Nauka, Moscow (1968).
4. F. F. Voronov, "Compression of bodies under high pressures," *Dokl. Akad. Nauk SSSR*, 226, No. 5 (1976).
5. E. G. Brovman, Yu. Kagan, and A. Kholas, "Properties of alkali metals," *Fiz. Tverd. Tela*, 12, No. 4, 1001 (1970).
6. V. G. Vak and A. V. Trefilov, "On a theory of the atomic properties of the alkali metals," *Fiz. Tverd. Tela*, 19, No. 1, 244 (1977).
7. E. V. Zarochentsev, V. P. Safronov, and A. V. Trefilov, "Third-order elastic moduli of the alkali metals," *Fiz. Nizk. Temp.*, 3, No. 2, 209 (1977).
8. Y. K. Huang, "Analytical approach to the shock compressibility of 18 cubic-lattice metals," *J. Chem. Phys.*, 53, No. 2, 571 (1970).
9. Ya. B. Zel'dovich and Yu. P. Raizer, *Physics of Shock Waves and High Temperature Hydrodynamic Phenomena*, Vols. 1 and 2, Academic Press, New York (1968).
10. F. A. Baum, L. P. Orlenko, K. P. Stanyukovich, V. P. Chelyshev, and B. I. Shekhter, *The Physics of an Explosion* [in Russian], Nauka, Moscow (1975).
11. L. D. Landau and E. M. Lifshits, *Statistical Physics*, Addison-Wesley, Reading, Massachusetts (1968).
12. N. N. Makarenko, A. M. Nikolaenko, V. A. Ivanov, and S. M. Stishov, "Equation of state of alkali metals: sodium," *Zh. Eksp. Teor. Fiz.*, 69, No. 5, 1723 (1975).
13. R. J. Becroft and C. A. Swenson, "An experimental equation of state for sodium," *J. Phys. Chem. Solids*, 18, No. 4, 329 (1961).
14. C. E. Monfort III and C. A. Swenson, "An experimental equation of state for potassium metal," *J. Phys. Chem. Solids*, 26, No. 2, 291 (1965).
15. S. N. Vaidya and G. N. Kennedy, "Compressibility of 22 elemental solids to 45 kbar," *J. Phys. Chem. Solids*, 33, No. 7, 1377 (1972).